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REVIEW Matter and Change SECTION 3
SHORT ANSWER Answer the following
questions in the space provided.

Modern Chemistry Chapter 1 Review Answers

HOLT MODERN CHEMISTRY -- CHAPTER 1

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REVIEW The following pages contain the bulk (but not all) of the information for the chapter 1 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test.

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CHAPTER 1 REVIEW Matter and Change

MIXED REVIEW SHORT ANSWER Answer

the following questions in the space

provided. 1. Classify each of the

following as a homogeneous or

heterogeneous substance.

homogeneous a. sugar homogeneous d.

plastic wrap homogeneous b. iron filings

heterogeneous e. cement sidewalk

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heterogeneous c. granola bar 2. For each type of investigation, select the most appropriate ...

1 Matter and Change - HUBBARD'S CHEMISTRY

CHAPTER 5 REVIEW The Periodic Law
SECTION 1 SHORT ANSWER Answer the following questions in the space

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provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

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5 The Periodic Law

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The element boron, B, has an atomic mass of 10.81 amu according to the periodic table.

3 Atoms: The Building Blocks of

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Matter

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(Section 7.1) Review Skills The presentation of information in this chapter assumes that you can already perform the tasks listed below. You can test your readiness to proceed by answering the Review Questions at the

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end of the chapter. This might also be a good time to read the Chapter Objectives, which precede the Review Questions.

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CHAPTER 9 REVIEW Stoichiometry

MIXED REVIEW SHORT ANSWER Answer

the following questions in the space provided. 1. Given the following

equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO$

$2(g) + 2H_2O(g)$ a. What is the value of the coefficient x in this equation? 40.07

g/mol b. What is the molar mass of C_3H_4 ? 2 mol O_2 :1 mol H_2O c. What is the

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mole ratio ...

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